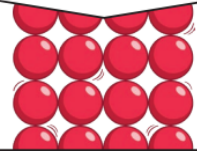
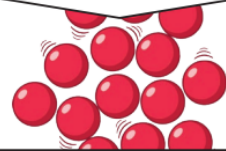

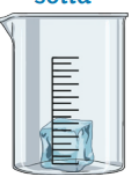


| Key Vocabulary | |
|-------------------------|--|
| states of matter | Materials can be one of three states: solids , liquids or gases . Some materials can change from one state to another and back again. |
| solids | These are materials that keep their shape unless a force is applied to them. They can be hard, soft or even squashy. Solids take up the same amount of space no matter what has happened to them. |
| liquids | Liquids take the shape of their container. They can change shape but do not change the amount of space they take up. They can flow or be poured. |
| gases | Gases can spread out to completely fill the container or room they are in. They do not have any fixed shape but they do have a mass. |
| water vapour | This is water that takes the form of a gas . When water is boiled, it evaporates into a water vapour . |
| melt | This is when a solid changes to a liquid . |
| freeze | Liquid turns to a solid during the freezing process. |
| evaporate | Turn a liquid into a gas . |
| condense | Turn a gas into a liquid . |
| precipitation | Liquid or solid particles that fall from a cloud as rain, sleet, hail or snow. |


| Key Knowledge | | |
|--|--|--|
| There are three states of matter. | | |
| Solid | Liquid | Gas |
|  |  |  |
| Particles in a solid are close together and cannot move. They can only vibrate. | Particles in a liquid are close together but can move around each other easily. | Particles in a gas are spread out and can move around very quickly in all directions. |

When water and other **liquids** reach a certain temperature, they change state into a **solid** or a **gas**. The temperatures that these changes happen at are called the boiling, **melting** or **freezing** point.


solid



heat

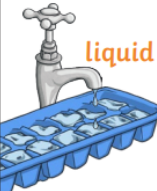


liquid

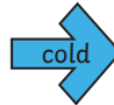


If a **solid** is heated to its **melting** point, it **melts** and changes to a **liquid**. This is because the particles start to move faster and faster until they are able to move over and around each other.


liquid



cold



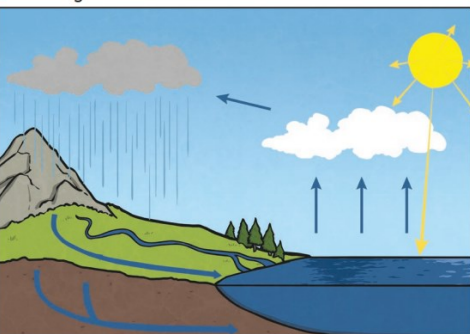
solid



When **freezing** occurs, the particles in the **liquid** begin to slow down as they get colder and colder. They can then only move gently on the spot, giving them a **solid** structure.

1. Water from lakes, puddles, rivers and seas is **evaporated** by the sun's heat, turning it into **water vapour**.
2. This **water vapour** rises, then cools down to form water droplets in clouds (**condensation**).
3. When the droplets get too heavy, they fall back to the earth as rain, sleet, hail or snow (**precipitation**).

Condensation and evaporation occur within the water cycle.



Evaporation



Evaporation occurs when water turns into **water vapour**. This happens very quickly when the water is hot, like in a kettle, but it can also happen slowly, like a puddle **evaporating** in the warm air.

Condensation



Condensation is when **water vapour** is cooled down and turns into water. You can see this when droplets of water form on a window. The **water vapour** in the air cools when it touches the cold surface.